

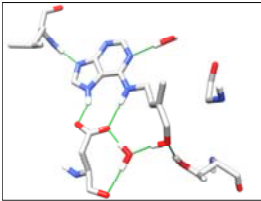
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## The Long and Rocky Road from a PDB File to a Protein-Ligand Docking Score

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**Stefan Bietz**  
**Nadine Schneider**  
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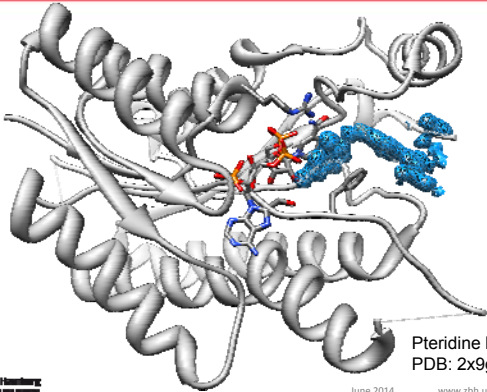


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## Protein Structures: The Starting Point for New Drugs

2



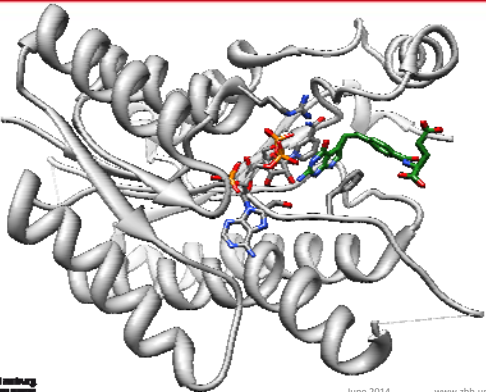
Pteridine Reductase  
PDB: 2x9g

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## Protein Structures: The Starting Point for New Drugs

3



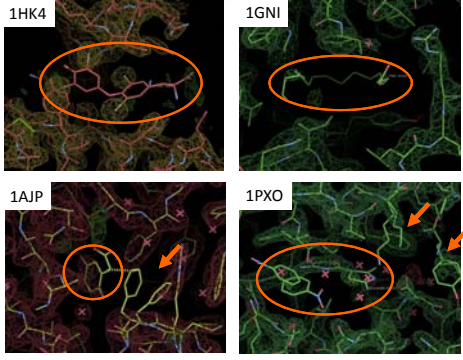
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## Detailed Structural Analysis of PDBbind 2007 by Consulting Electron Density Maps

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- 107 complexes with structural deficiencies
- Missing electron density
- Alternative conformations
- High temperature factors
- Contact with symmetry related subunits



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### Protein Structures: The Starting Point for New Drugs

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### Protein Structures: The Starting Point for New Drugs

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### Does it Matter?

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$\Delta G = -28 \text{ kJ/mol}$

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$\Delta G = -34 \text{ kJ/mol}$

\* Foloppe N. et al., JMedChem 2005

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### Contribution of a Single Hydrogen Bond to Binding Affinity

- Example: Thrombin/Trypsin-inhibitor complexes
- Exchange NH  $\rightarrow$  CH<sub>2</sub>:
  - slight loss in IC<sub>50</sub>
- Exchange NH  $\rightarrow$  O:
  - loss of two hydrogen bonds upon dehydration
  - no formation of a new hydrogen bond
  - IC<sub>50</sub> drops by about three orders of magnitude

Abb.: G. Klebe, Wirkstoffdesign Spektrum Akad. Verlag, 2009

Tabelle 4.4 Bindung von 4.4 an die Serinproteasen Thrombin bzw. Trypsin

Enzym	IC <sub>50</sub> -Werte in mg/ml X = -NH-	-O-	-CH <sub>2</sub> -
Thrombin	0,009	52	0,07
Trypsin	0,009	43	0,018

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Urbaczek et al, JCIM (2013), 53, 76-87

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### PHASE 1: Perception of Chemical Structure

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Identification of Bonds → Selection & Scoring of Valence States → Generation of Valence Bond Forms → Scoring of Valence Bond Forms

Atomic Coordinates → Best Solution

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Urbaczek et al, JCIM (2013), 53, 76-87

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### Valence States

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- Valence state: Chemically valid combination of bond orders and formal charge for a particular element.
- Aim: Find a valid valence state combination (VSC)

number of bonds single / double / triple  
Element → N210+ ← Formal Charge

Forms of Histidine A, Forms of Chinon B, Kekule Structures C

Legend:  
Tautomers (Red)  
Protonation States (Green)  
Resonance Forms (Blue)  
Kekule Structures (Orange)  
Redox Forms (Cyan)

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Urbaczek et al, JCIM (2013), 53, 76-87

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### Valence States

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- Atoms get sets of all valence states compatible to the detected bond structure
- Each valence state is scored by the geometry of surrounding covalent bonds

Bond Angle ( $\alpha$ ), Torsion Angle ( $\tau$ ), Triple Product ( $n$ )

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Urbaczek et al, JCIM (2013), 53, 76-87

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### From Valence States to Valence Bond Structures

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- Valence states get probabilities converted to scores
- Iterative refinement by removing valence states incompatible to those of neighboring atoms
- Partitioning of the molecules into zones of covalently bound atoms with yet undefined valence state
- Branch&Bound procedure enumerating and selecting highest scoring valence states
- Consider aromaticity, favorize certain functional group topologies in case of nearly equal scores

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Urbaczek et al, JCM (2013), 53, 76-87

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### Results: Quality of Chemical Structure Perception

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- 363 PDB files with hand-curated structures, 563 ligands in total
- NAOMI vs. Ligand Expo:
  - 92% identical structures
  - differences found: (NAOMI structure validated by literature):

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### PHASE 2: Prediction of Protonation

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### Prediction of Hydrogen Coordinates

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- Existing hydrogen placing tools:
  - WHAT IF<sup>1</sup>
  - Protonate3D (MOE)<sup>2</sup>
  - HINT Comp. Titration<sup>3</sup>
  - Protoss<sup>4</sup>
  - YASARA<sup>5</sup>

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<sup>1</sup> R.W.W. Hooft et al. Proteins, 26(4):363-376, 1996.  
<sup>2</sup> P. Labute et al. Proteins, 75(1):187, 2009.  
<sup>3</sup> A.S. Bayden et al. Journal of Computer-Aided Molecular Design, 23(9): 621-632, 2009.  
<sup>4</sup> T. Lippert et al. Journal of Cheminformatics, 1(1):13, 2009.  
<sup>5</sup> E. Krieger et al. Computational Drug Discovery and Design, 8:19-405-421, 2012.

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### Prediction of Hydrogen Coordinates

16

	proteins					ligands			
	rotatable hydrogens	tautomers	protonation states	side chain flips	water orientations	rotatable hydrogens	tautomers	protonation states	side chain flips
WHAT IF	✓	✓	✓	✓	✓				
Protonate3D	✓	✓	✓	✓	✓	✓	✓		
HINT CT	✓	✓	✓	✓	✓	✓	✓	✓	
Protoss v2009	✓	✓	✓	✓	✓	✓	✓	✓	
YASARA	✓	✓	✓	✓	✓	✓	✓		✓

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## Error Case Examples

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3k3i      1j9      1y6t

Repulsive donor interactions      Hydrogen-metal clashes      Unsaturated donors / acceptors

YASARA      Protonate3D      Protoss v2009

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## Frequency of Tautomerism

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- Ligand Expo<sup>6</sup>: 17,563 small molecules<sup>7</sup>
- According to the ProToss protonation model
  - 81 % exhibit alternative tautomers or / and protonation states
  - 17 % only contain commonly used groups (primary amines, carboxylates, imidazoles)
  - Only 19 % do not show variability
  - 1802 structurally different variable mode regions (VMRs)

portion of molecules [%]

ensemble size

different VMR type portions

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## Challenges to an Advancement of ProToss

Bietz et al, JCheminf (2014), 6 (12):1-12

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- Find the tautomer / protonation state that leads to the best hydrogen bonding network in the protein-ligand complex
- Keep the runtime low

	proteins				ligands			
	variable hydrogen backbone	protonation states	side chain flip	water molecules	variable hydrogen backbone	protonation states	side chain flip	
WHAT IF	✓	✓	✓	✓	✓	✓	✓	
Protonate3D	✓	✓	✓	✓	✓	✓	✓	
HINT CT	✓	✓	✓	✓	✓	✓	✓	
Protona v2009	✓	✓	✓	✓	✓	✓	✓	
YASARA	✓	✓	✓	✓	✓	✓	✓	
Protoss v2013	✓	✓	✓	✓	✓	✓	✓	✓

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## Protoss Workflow

Bietz et al, JCheminf (2014), 6 (12):1-12

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variable groups

scoring

hydrogen-bonding network

dynamic programming algorithm

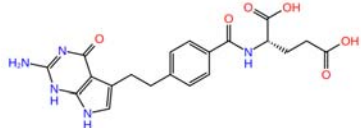
optimal network

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## Generation of Tautomers and Protonation States

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Pemetrexed (PTR1 inhibitor)

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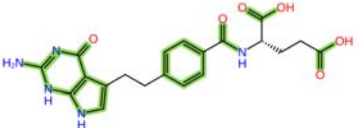
Urbaczek et al, J Chem Inf Model (2014), 45 (3):756-766

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## Generation of Tautomers and Protonation States

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Step 1: Conjugated zone identification



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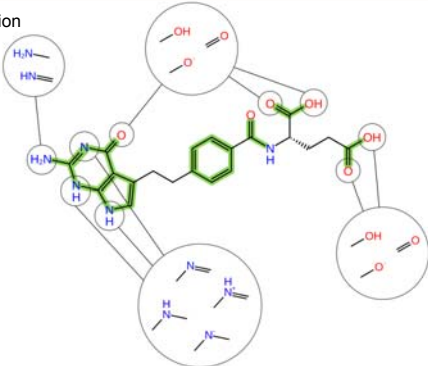
Urbaczek et al, J Chem Inf Model (2014), 45 (3):756-766

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## Generation of Tautomers and Protonation States

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Step 2: Atom state selection



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Urbaczek et al, J Chem Inf Model (2014), 45 (3):756-766

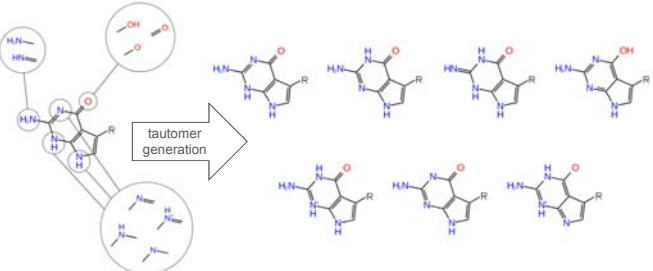
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## Generation of Tautomers and Protonation States

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Step 3: Enumeration of tautomers and protonation states

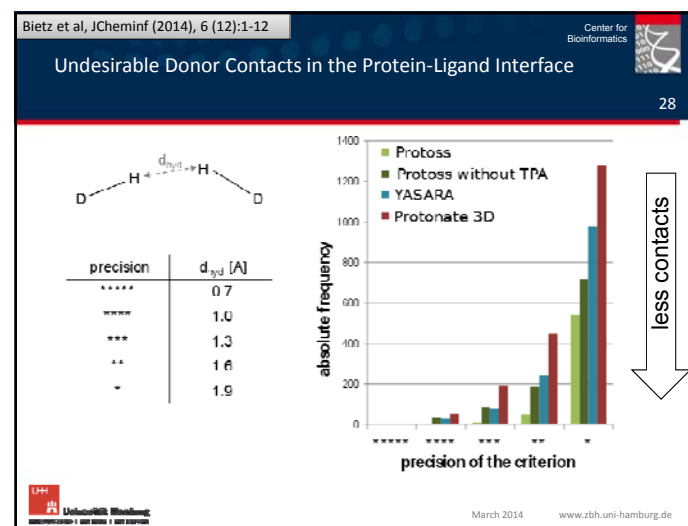
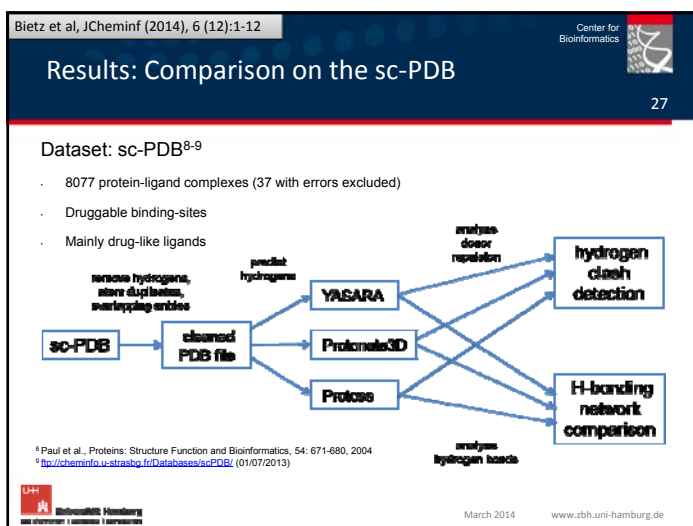
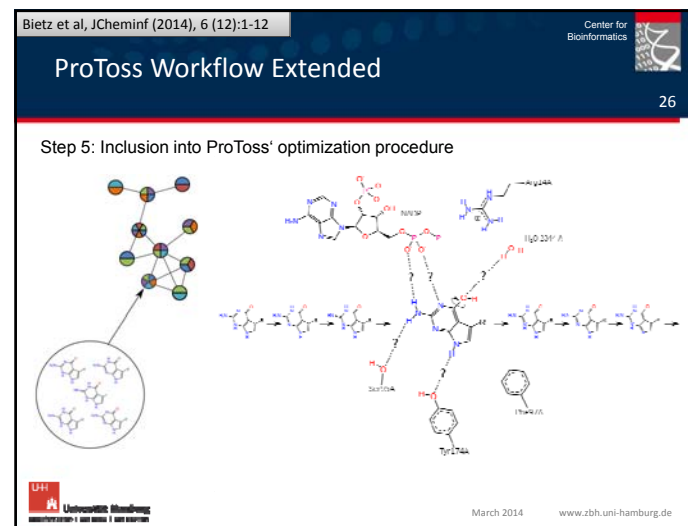
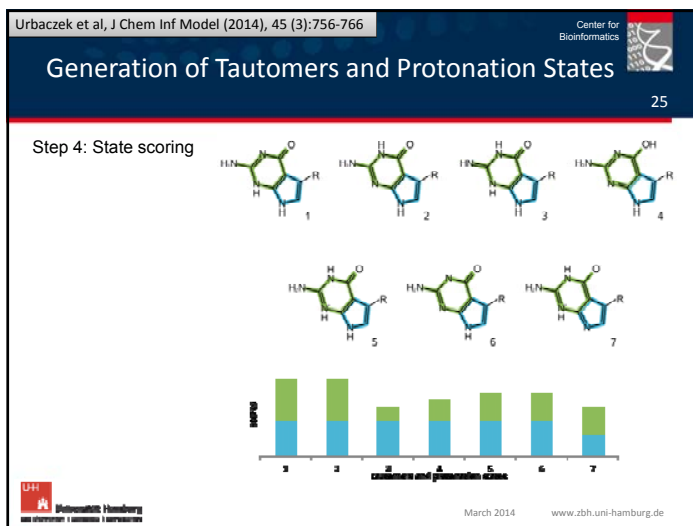
tautomer generation

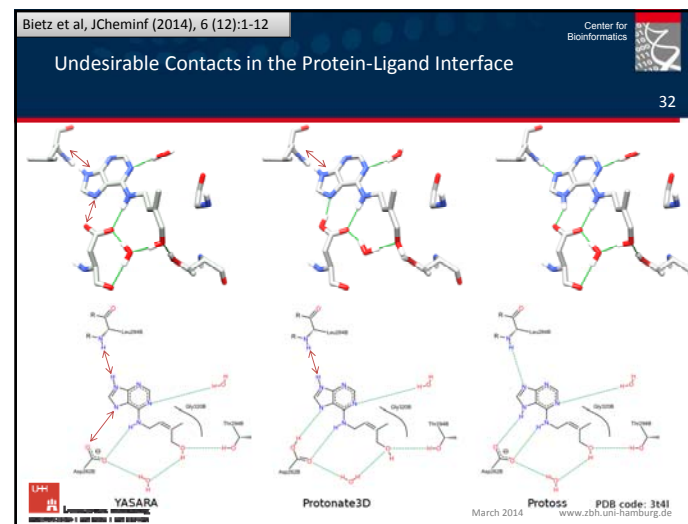
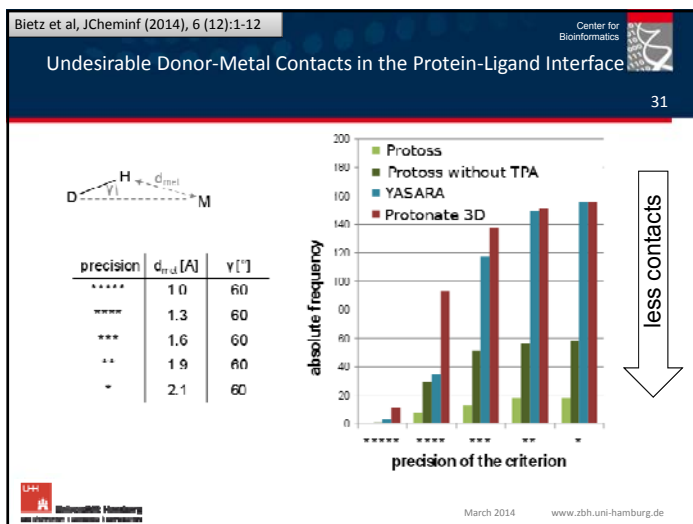
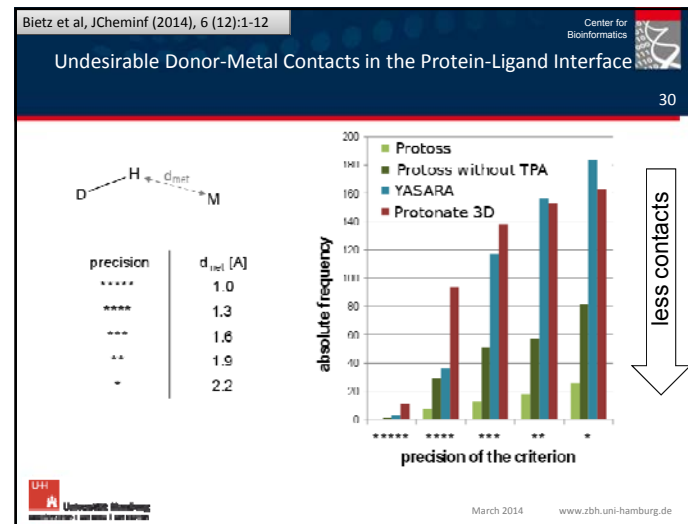
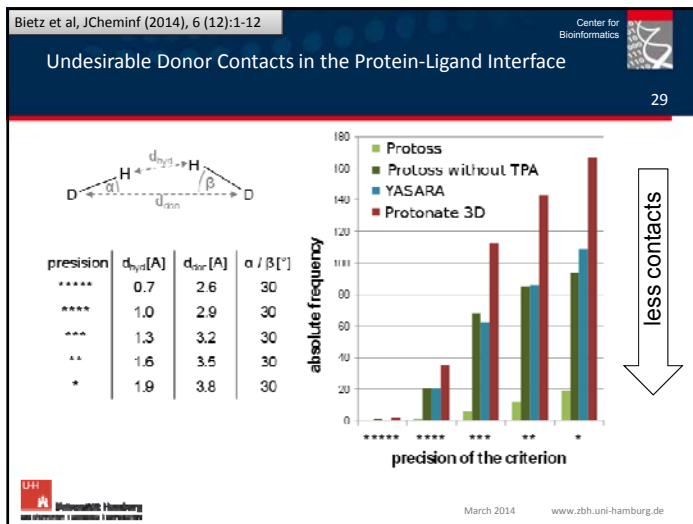


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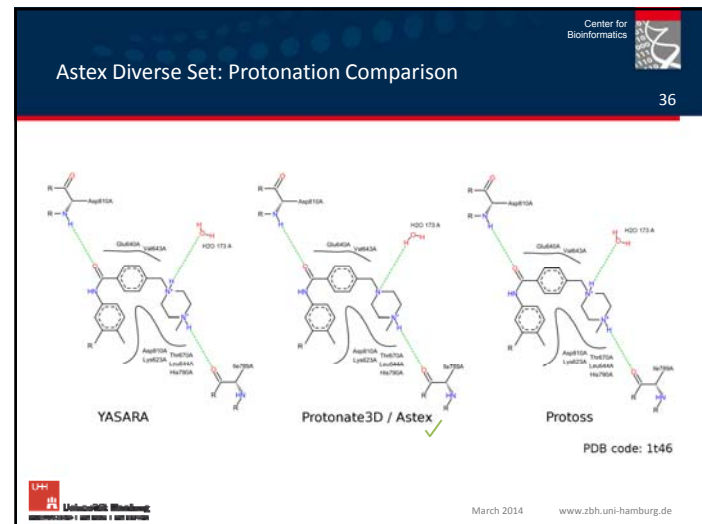
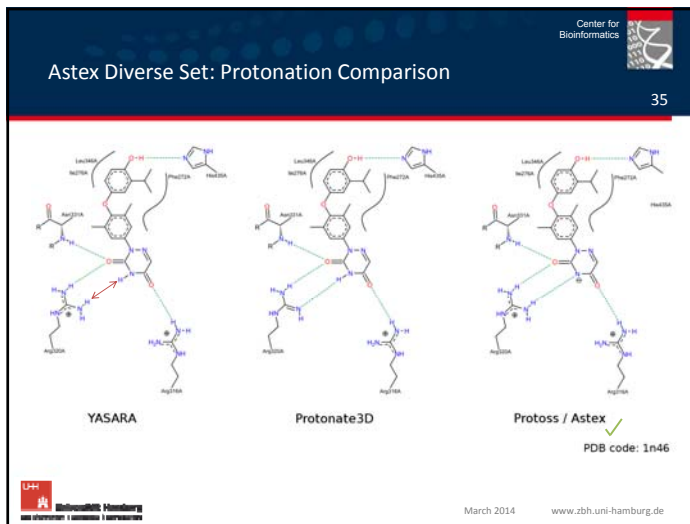
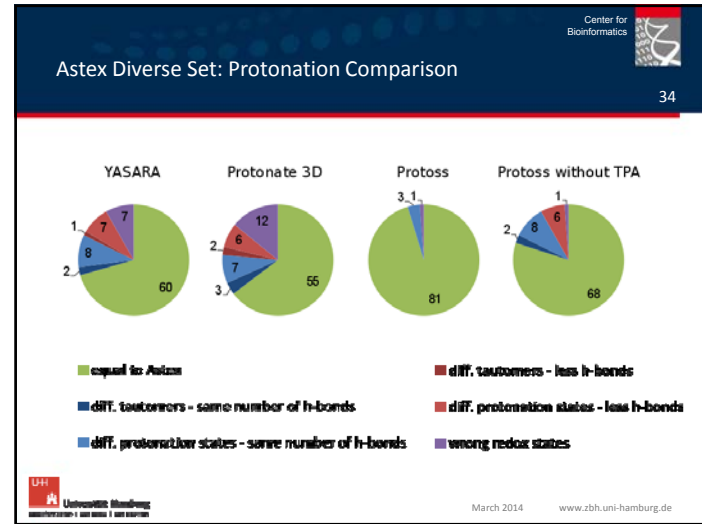
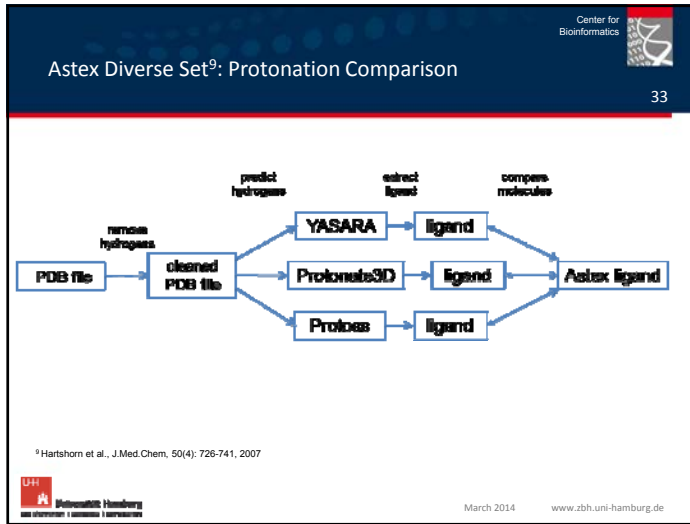
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




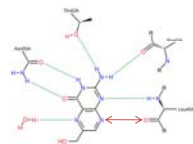
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## Astex Diverse Set: Protonation Comparison

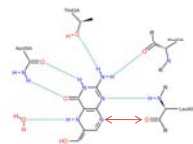
37



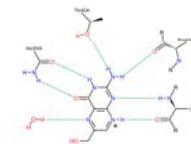
6-hydroxymethyl-7,8-dihydropterin



YASARA




Protonate3D



Protoss

PDB code: 1hq2




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
## Astex Diverse Set: Protonation Comparison

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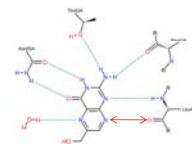
6-hydroxymethyl-7,8-dihydropterin (1hq2)

1.35 Å

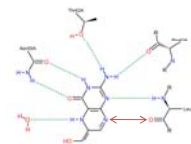


6-hydroxymethylpterin (3lp0)

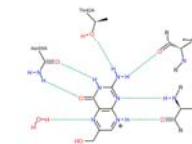
1.34 Å



YASARA




Protonate3D



Protoss

PDB code: 1hq2



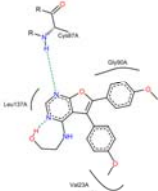
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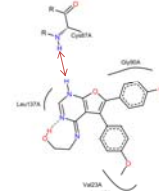
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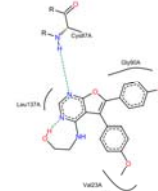
39



YASARA / Astex ✓




Protonate3D



Protoss / Astex ✓

PDB code: 2br1



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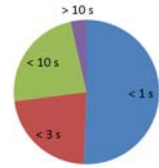
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## ProToss Runtime Evaluation

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- Whole complex optimization on the Astex Diverse Set (including all ligands, co-factors and water molecules)
- Runtimes include
  - File IO
  - Complex preprocessing
  - Network optimization
- Intel® Core™ i7-2600 CPU ( 3.4 GHz ) and 8 GB memory
- Mean: 2.47 s
- Median 0.93 s





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PHASE 3: Scoring Protein-Ligand Complexes

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Schneider et al, *J. Comput. Aided Mol. Des.*, **27**(1), 15-29 (2013)

HYDE – Towards a Consistent Description of Hydrogen Bonding, Dehydration and the Hydrophobic Effect

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$$\Delta G_{HYDE} = \sum_{atoms\ i} \Delta G_{dehydration}^i + \Delta G_{H-bond}^i$$

$$\Delta G_{dehydration}^i = -2.3 RT \cdot plogP^i \cdot \Delta acc^i$$

$$\Delta G_{H-bond}^i = \frac{2.3 RT}{F_{sat}} \cdot plogP^i \cdot \Delta sat^i$$

unfavorable energy

favorable energy

1GKC

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Schneider et al, *J. Comput. Aided Mol. Des.* **26** 2011: 701-723

HYDE Benchmarks

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- PDBbind 2007 coreset<sup>[1]</sup> → Binding affinity prediction
- Astex diverse set<sup>[2]</sup> → Redocking experiments
- Dataset of Useful Decoy<sup>[3]</sup> → Virtual screening study

[1] Wang R. et al, *JMedChem*, 2004, 2005  
 [2] Hartshorn M. J. et al, *JMedChem*, 2007  
 [3] Huang N. et al, *JMedChem*, 2006  
 [4] Schneider N. et al, *JCAMD*, 2011

Mean AUC	0.77
STD	0.17
Median AUC	0.78
Min AUC	0.50
Max AUC	0.95

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Estimating the Cost of a Single Hydrogen Bond

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$\Delta G = -28\text{ kJ/mol}$

$\Delta G_{HYDE} = -22\text{ kJ/mol}$

$\Delta G_{HYDE} = -32\text{ kJ/mol}$

$\Delta G = -34\text{ kJ/mol}$

2BRB

\* Foloppe N. et al., *JMedChem* 2005

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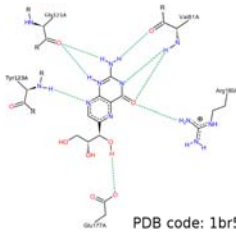
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## A Few Things to Remember

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- PDB files are models – consult electron density.
- Perception of chemical structures from PDB is difficult – cross-check with literature.
- Assigning protonation and tautomeric states is not simpler
  - cross-check the results of automatic assignment procedures
- Tools like ProToss help in large scale applications in which manual curation is not possible.



PDB code: 1br5



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## Contributions / Acknowledgements


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AMD Group



- **Stefan Bietz**  
ProToss development
  - **Sascha Urbaczek**  
Tautomer generation
  - **Tobias Lippert**  
First version of ProToss
  - **Nadine Schneider**  
Hyde development
- **Benjamin Schulz**  
ProToss development, testing and validation
  - **Holger Claussen**  
Testing, testing ...
  - **Christian Lemmen**





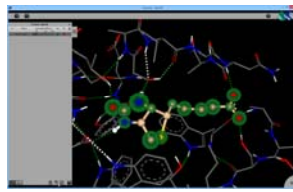
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## Software Availability


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More on ProToss  
<http://protoss.zbh.uni-hamburg.de/>  
 ProToss in action:  
<http://www.biosolveit.de/SeeSAR/>



More on tools and servers from ZBH:  
<http://www.zbh.uni-hamburg.de/AMD>  
<http://www.biosolveit.de/techtransfer>





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