

Chemoinformatics in Drug Discovery - Quo Vadis?

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Chemoinformatics, June 20-24, 2010

| | |
|--|---|
| $C_2H_4O_2$ | empirische Formel. |
| $C_2H_4O_2 + H_2O$ | zusätzliche Formel. |
| $C_2H_4O_2 + H_2O$ | Wasserstofftheorie. |
| $C_2H_4O_2 + O_2$ | Kavaliertheorie. |
| $C_2H_4O_2 + 2H_2$ | Langechamp's Annahme. |
| $C_2H_4O_2 + H_2O$ | Brennbarer Ansatz. |
| $C_2H_4O_2 \cdot H_2O$ | Radikaltheorie. |
| $C_2H_4O_2 + O_2 + H_2O$ | Radikalketten. |
| $C_2H_4O_2 \xrightarrow{H_2O} C_2H_4$ | Gerhardt's Typentheorie. |
| $C_2H_4O_2$ | Typentheorie (Schleicher und Leibiger). |
| $C_2O_4 + C_2H_4 + H_2O$ | Benzaline's Pseudotheorie. |
| $H_2O \cdot (C_2H_4O_2 \cdot H_2O)$ | Kohler's Ansatz. |
| $H_2O \cdot (C_2H_4O_2 \cdot H_2O) + H_2O$ | BBK. |
| $C_2(H_2O)O_2$ | Wurtz. |
| $C_2H_4O_2(C_2H_4O_2)O_2$ | Brandstetter. |
| $C_2H_4O_2(H_2O)C_2H_4O_2$ | Gmelin. |
| $\frac{1}{2}(C_2H_4O_2)^2 + H_2O$ | Roschdorff. |
| $(C_2H_4O_2 + CO_2) + H_2O$ | Zerrini. |
| $C_2H_4O_2$ | Witt. |

Historical Formulas of Acetic Acid (1860)

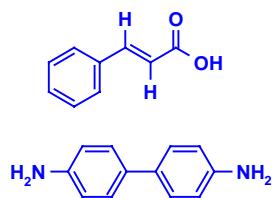
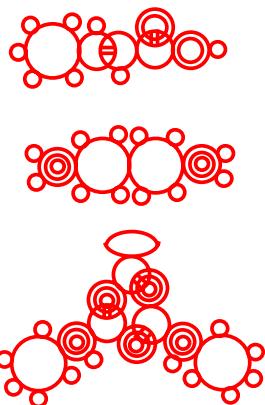


Joseph Loschmidt

"Chemische Studien. A. Constitutions-
Formeln der organischen Chemie in
geographischer Darstellung", Wien, 1861



Loschmidt Constitution Formulas (1861)



Kekulé Benzene Formula (1865)



QSAR



Corwin Hansch
(* 1918)

(picture taken at the
5th EuroQSAR, 1984)

S. L. Carney (DDT 9, 158-160 (2004)):
Has there been a single development
that, in your opinion, has moved the
field of medicinal chemistry ahead
more than any other?

Robin Ganellin (Professor of Medicinal
Chemistry, University College, London,
UK): I would go back to the 1960s to
the work of Corwin Hansch on the
importance of lipophilicity. ... that
changed the way of thinking in medi-
cinal chemistry. the application of
physical organic chemical approaches
to structure-activity analysis [has]
been very important.

Is QSAR relevant to Drug Discovery?

A. M. Doweyko, *Idrugs* 11, 894-899 (2008)

QSAR: dead or alive?

A. M. Doweyko, *J. Comput.-Aided Mol. Design* 22, 81-89 (2008)

On outliers and activity cliffs - why QSAR often disappoints

G. M. Maggiora, *J. Chem. Inf. Model.* 46, 1535 (2006)

Beware of q^2 !

A. Golbraikh and A. Tropsha, *J. Mol. Graphics & Model.* 20, 269-276(2002)

3D-QSAR illusions

A. M. Doweyko, *J. Comput.-Aided Mol. Design* 18, 587-596 (2004)

The trouble with QSAR (or how I learned to stop worrying and embrace fallacy)

S. R. Johnson, *J. Chem. Inf. Model.* 48, 25-26 (2008)

How not to develop a QSAR/QSPR relationship

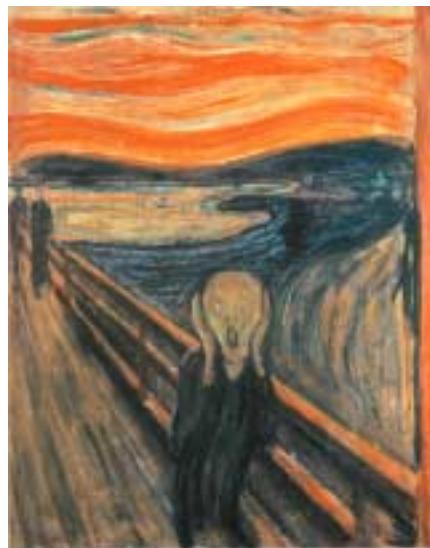
J. C. Dearden et al., *SAR and QSAR in Environ. Res.* 20, 241-266 (2009)

How to recognize and workaround pitfalls in QSAR studies: a critical review

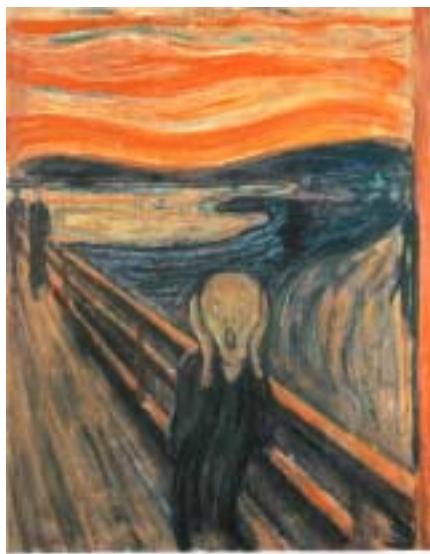
T. Scior et al., *Curr. Med. Chem.* 16, 4297-4313 (2009)

QSAR: Problems in Statistical Analyses

- inappropriate biological data
- wrong scaling of biological data
- data from different labs
- different binding modes
- mixed data (e.g. oral absorption and bioavailability)
- different mechanism of action (e.g. toxicity data)
- too few data points
- too many single points
- lack of chemical variation
- clustered data
- small variance of y values
- systematic error/s in y
- too large errors in y values
- outliers / wrong values
- wrong model selection



QSAR: Problems in Statistical Analyses



inappropriate x variables
too many x variables (Topliss)
a) in the model selection
b) in the final model
x variable scaling in CoMFA fields
interrelated x variables
singular matrix
elimination of variables that are
significant only with others
insignificant model (F test)
insignificant x variables (t test)
no qualitative (biophysical) model
no causal relationship (the storks)
extrapolation too far outside of
observation space
no validation method applied
wrong validation method,

How the Trouble Started: Connectivity Indices ${}^i\chi$

Connectivity indices
= electron-weighted
subgraph counts

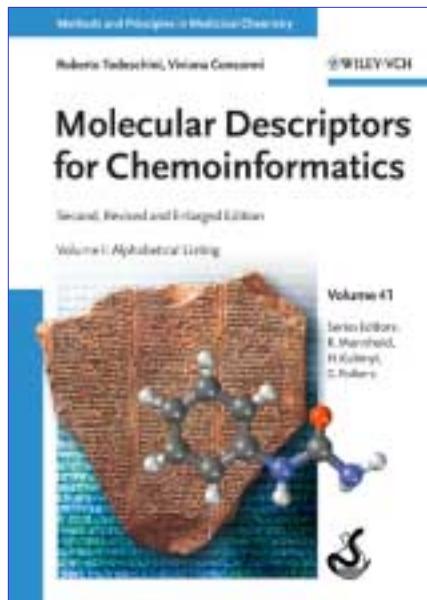
$${}^0\chi = \Sigma (\# \sigma\text{-electrons of } i)^{-0.5}$$

$${}^1\chi = \Sigma ({}^0\chi(i) \cdot {}^0\chi(j))^{-0.5} \quad (\text{over all bonds } ij)$$

... etc.



${}^0\chi \quad {}^1\chi \quad {}^2\chi \quad {}^3\chi_P, {}^3\chi_C \dots$



Program E-DRAGON

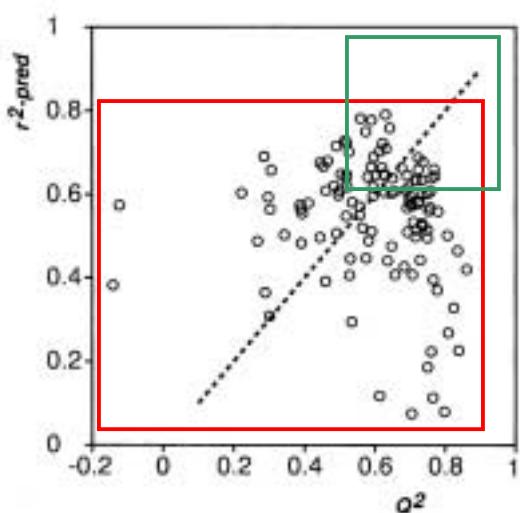
Roberto Todeschini

Web version of the DRAGON program at www.vcclab.org/lab/edragon/

E-DRAGON analyses up to 149 molecules and up to 150 atoms per molecule. Current version: Dragon 5.4 from March 28, 2006.

Calculates more than 1,600 molecular descriptors, organized in 20 blocks, from SMILES code, SDF, or MOL2 files.

External vs. Internal Predictivity

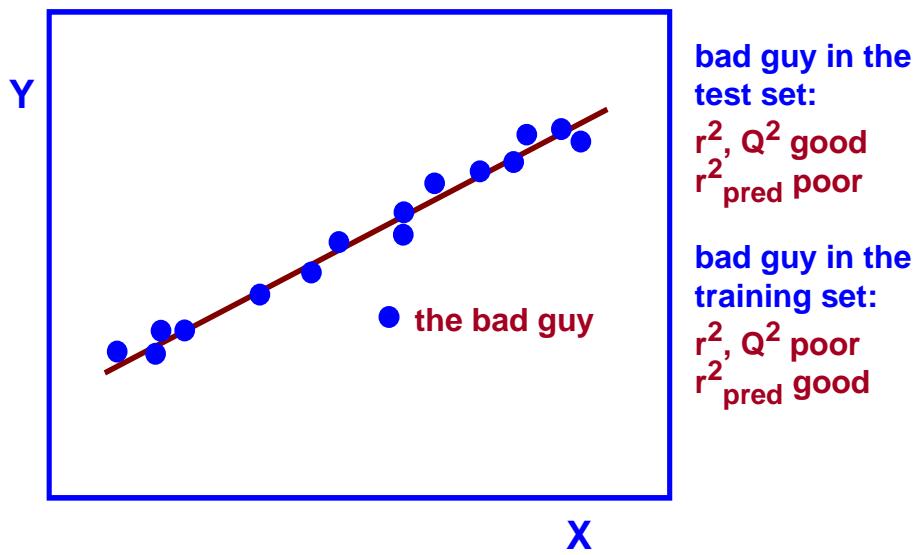


The „Kubinyi Paradox“

J. H. van Drie, Curr. Pharm. Des. 9, 1649-1664 (2003);
J. H. van Drie, in: Computational Medicinal Chemistry for Drug Discovery, P. Bultinck et al., Eds., Marcel Dekker, 2004, pp. 437-460.

Data from H. Kubinyi et al., J. Med. Chem. 41, 2553-2564 (1998).

„Good“ and „Bad“ Guys in Regression Analysis



Proper Validation of QSAR and 3D QSAR Models

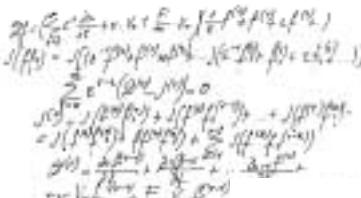
| Validation Method | Effect |
|---|--|
| Crossvalidation, using the original variables (LOO CV, LMO CV) | insufficient for model validation |
| Y scrambling, using the original variables | misleading |
| Y scrambling with new variable selection | may be misleading |
| Leave-one-out crossvalidation with new variable selection in every CV run | misleading in larger data sets |
| Leave-many-out (up to 30%) cross-validation with new variable selection in every CV run | the only reliable validation procedure |

see also T. Scior et al., Curr. Med. Chem. 16, 4297-4313 (2009)



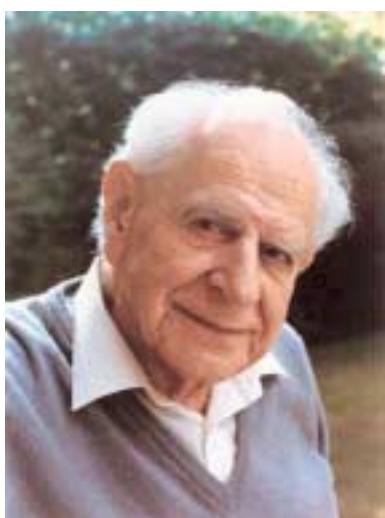
“Good” QSAR

- parameters with biophysical relevance
- few variables to select
- few variables in the model
- validation by LOO, LMO, y scrambling



“Poor” QSAR

- artificial parameters
- too many variables to select
- many variables in the model
- no test set predictivity (“Kubinyi paradox”)



Good and Poor Science

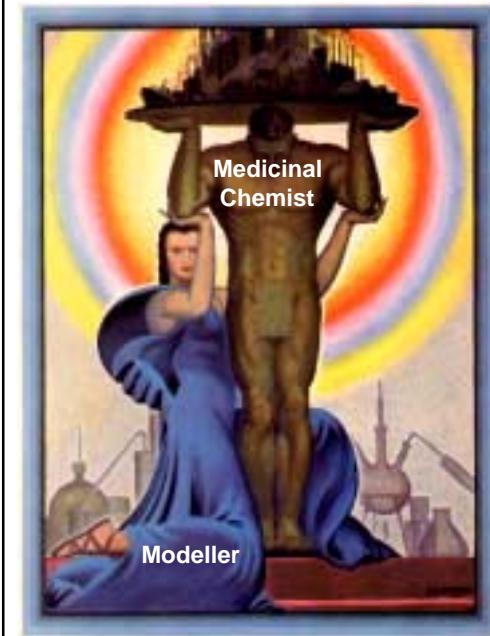
[one has to] „differentiate between science and pseudoscience, knowing very well that science often errs and that pseudoscience may happen to stumble on the truth“

„it is easy to obtain confirmations - if one looks for them“

„a theory which is not refutable ... is non-scientific“

„some theories, when found to be false, are still upheld by their admirers - for example by introducing some auxiliary assumption, or by reinterpreting the theory *ad hoc* in such a way that it escapes refutation“

Sir Karl Popper
★1902 Vienna, † 1998 London



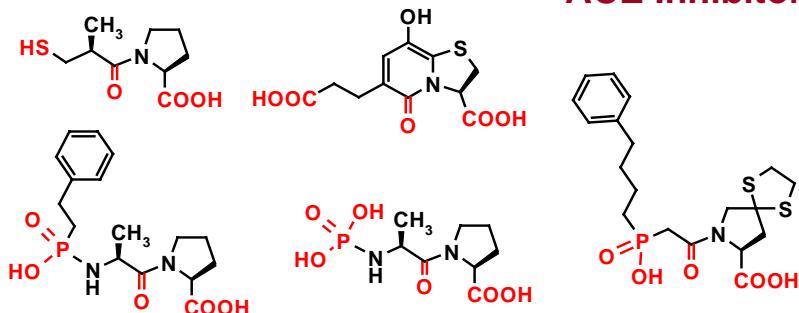
A. Cressy Morrison

**Man in a Chemical World
The Service of Chemical Industry**

Ch. Scribner's Sons, NY, 1937

„Chemical Industry, Upheld
by Pure Science, Sustains
the Production of Man's
Necessities“

Historical Pharmacophore Definition: ACE Inhibitors

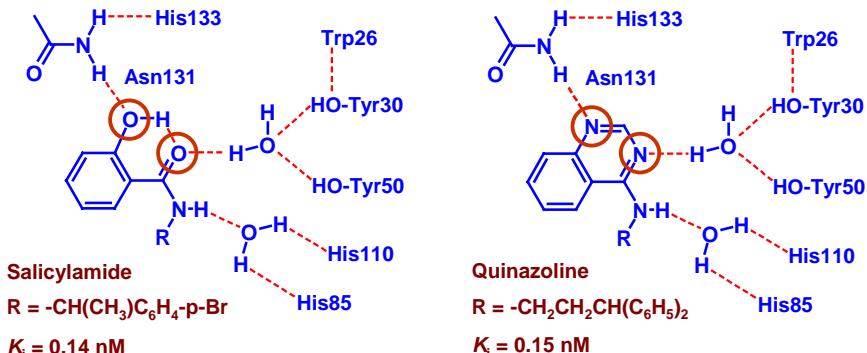


defined by
functional groups

-SH, -COOH,
-PO₃H₂, >PO₂H

C
= O
COOH

Receptors Just Recognize Properties



A pharmacophore is the ensemble of steric and electronic features that is necessary to ensure the optimal supramolecular interactions with a specific biological target and to trigger its biological response.

C. G. Wermuth et al., Pure Appl. Chem. 70, 1129-1143 (1998)

Pharmacophore Generation and 3D Searches

Catalyst (Accelrys)

established tool for hypothesis generation and 3D searches

CATS (Roche) topological pharmacophores - no 3D structures required

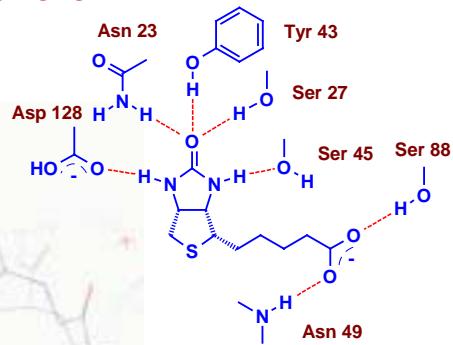
FTree (feature trees; BioSolveIT)

no 3D structures required, ultrafast searches

LigandScout (inte:ligand)

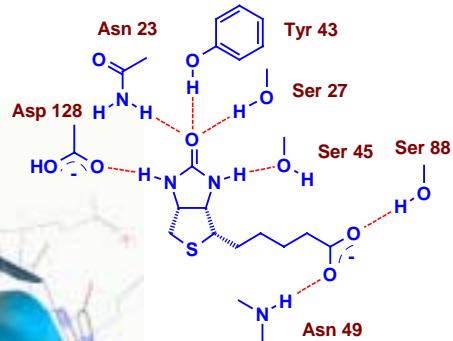
automated generation of pharmacophores from protein 3D structures; ligand-based pharmacophore generation; 3D searches

LigandScout Pharmacophore Recognition (inte:ligand)



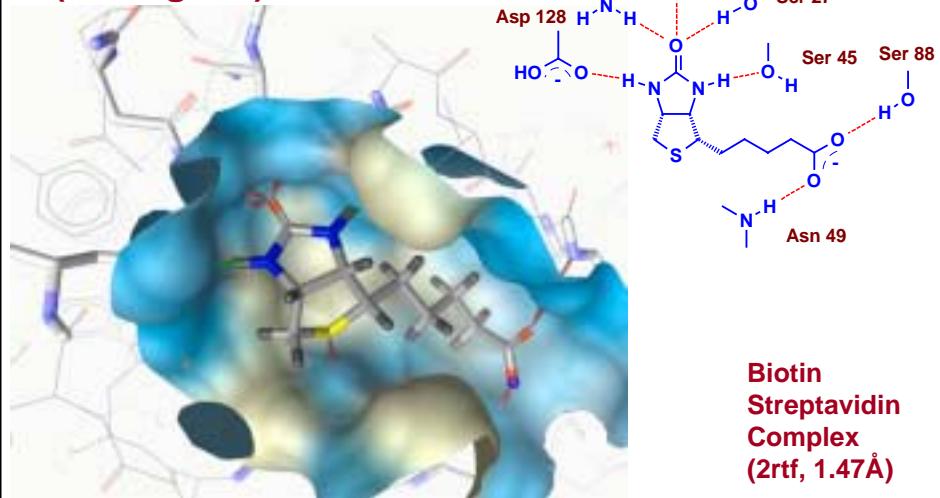
Biotin
Streptavidin
Complex
(2rtf, 1.47Å)

LigandScout Pharmacophore Recognition (inte:ligand)

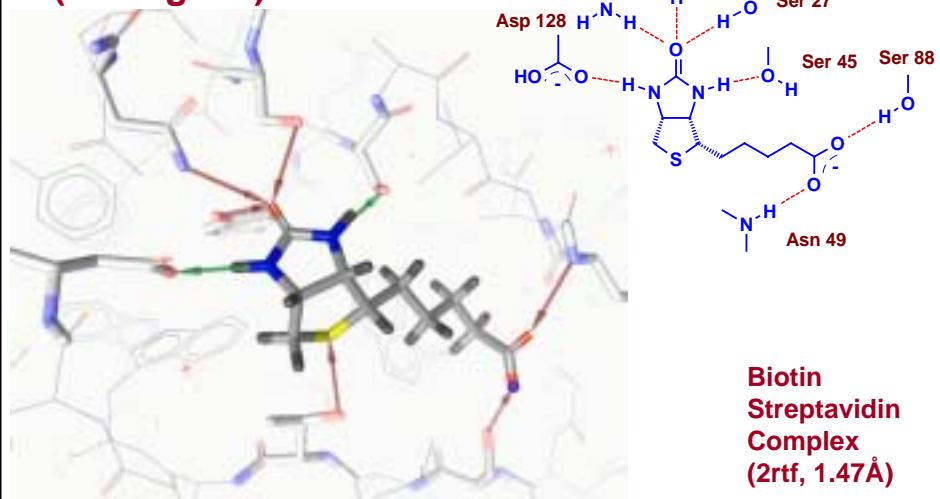


Biotin
Streptavidin
Complex
(2rtf, 1.47Å)

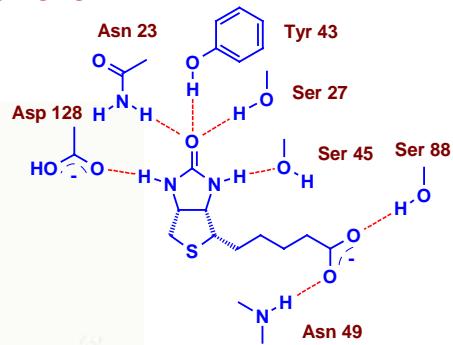
LigandScout Pharmacophore Recognition (inte:ligand)



LigandScout Pharmacophore Recognition (inte:ligand)

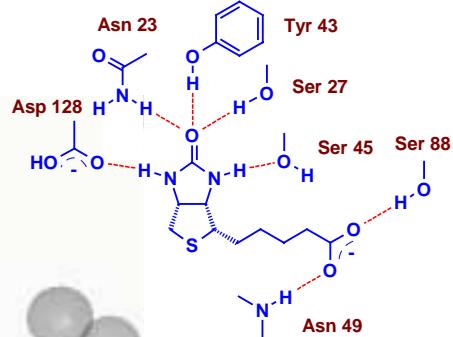


LigandScout Pharmacophore Recognition (inte:ligand)



Biotin
Streptavidin
Complex
(2rtf, 1.47Å)

LigandScout Pharmacophore Recognition (inte:ligand)



Biotin
Streptavidin
Complex
(2rtf, 1.47Å)

Problems in Pharmacophore Generation

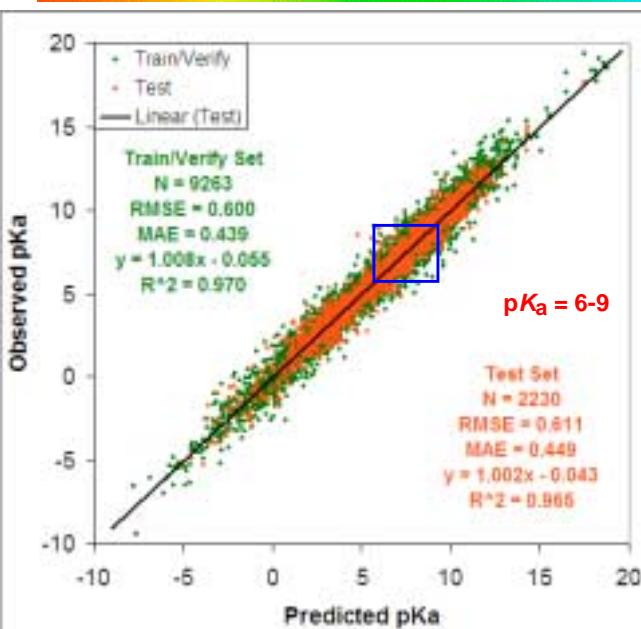
Isomers, enantiomers, diastereomers

Ionisation and Dissociation
(Sadowski rules, ACS Boston, 2002)

Tautomeric and protomeric forms
(program AGENT, ETH Zurich; ChemoSoft, ChemDiv;
LigPrep, Schroedinger; and several others)

Acceptor properties of oxygen and sulfur atoms
(esters, aromatic ethers, oxazoles,
isoxazoles, thiazoles, etc.)

Superposition of flexible molecules



Software
for pK_a
Prediction

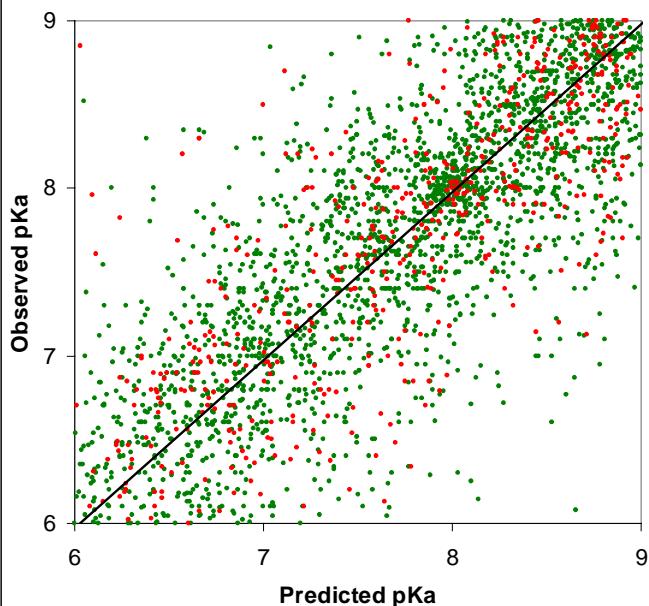
pK_a model in
ADMET Predictor
4.0

[www.simulations-
plus.com/Definitions.
aspx?ID=55](http://www.simulations-plus.com/Definitions.aspx?ID=55)

Software for pK_a Prediction

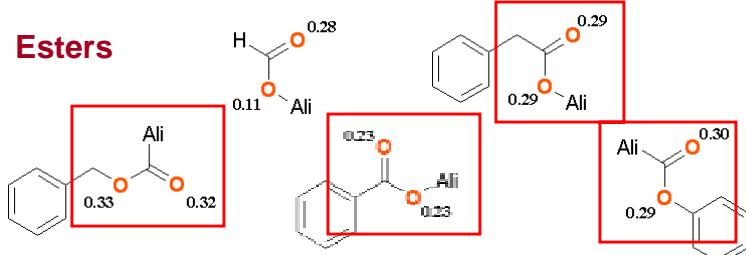
pK_a model in
ADMET Predictor
4.0

courtesy of Robert
Fraczkiewicz,
Simulations Plus, Inc.

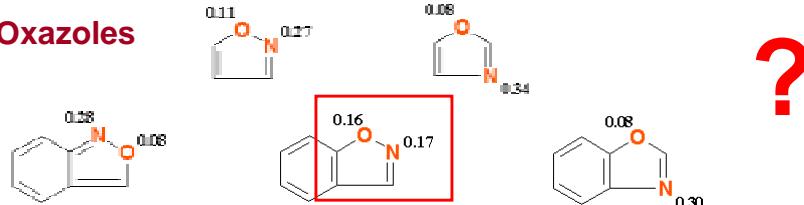


Acceptor Potentials of Esters and Oxazoles

Esters



Oxazoles

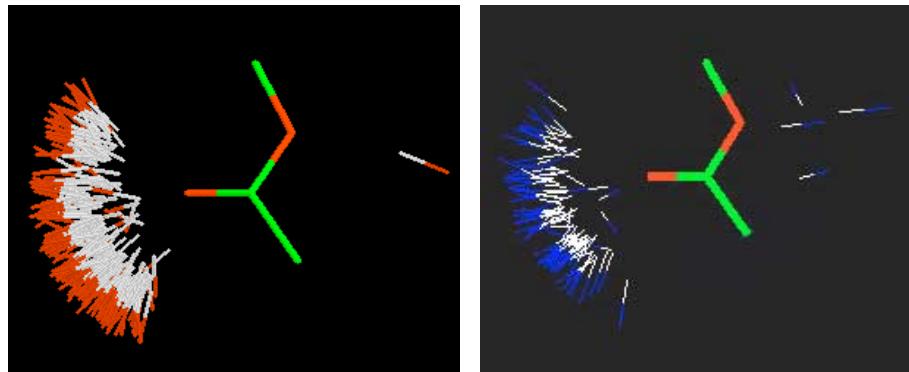


Pharmacophore Analyses Must Consider Correct Donor and Acceptor Properties of Ligands

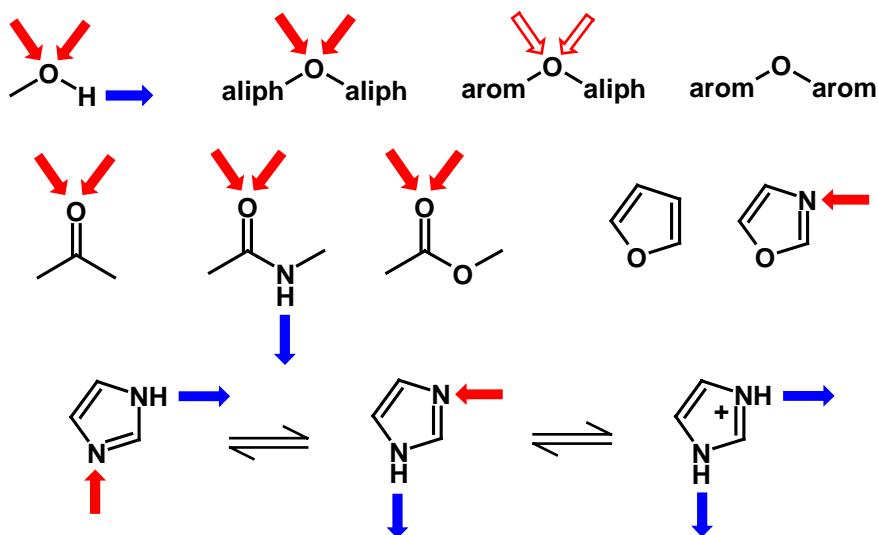
The billion dollar question:

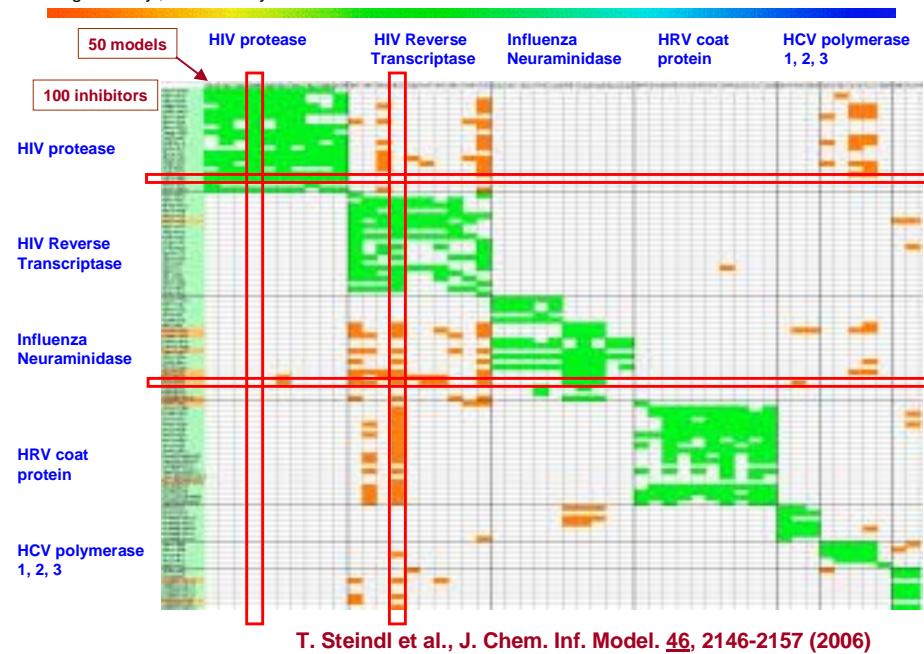
how many acceptor positions has an ester group ?

Correct answer: Two, but why?



Donor and Acceptor Properties of O and N



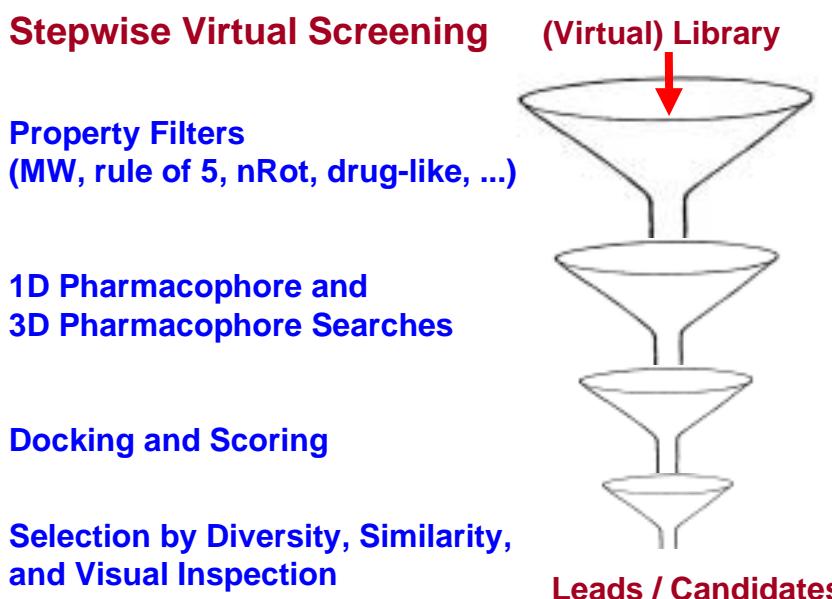


Drug Research is



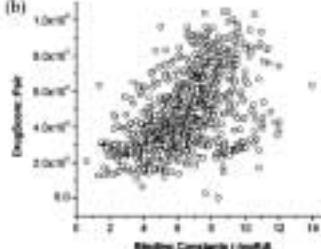
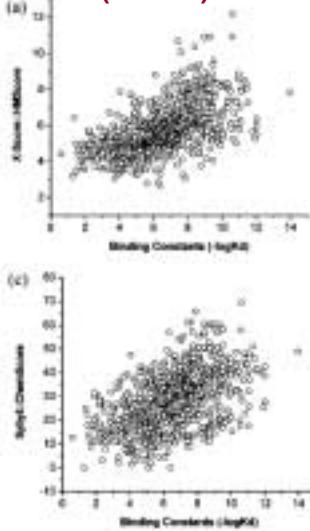
the Search for a Needle in a Haystack

| Tools for Virtual Screening | remaining |
|---|-----------|
| Garbage filter | 90% |
| Druglike / Non-druglike | 75% |
| Bioavailability | 60% |
| Cytotoxicity | : |
| hERG channel inhibiton | : |
| Antitargets | : |
| α_{1a} (orthostatic hypotension) | : |
| D2 (extrapyramidal syndrome) | : |
| 5-HT _{2c} (obesity) | : |
| musc. M1 (hallucinations, memory) | : |
| CYP inhibition (3A4, 2C9, 2D6) | : |
| Pharmacophore searches | : |
| Docking and scoring | 0% ? |

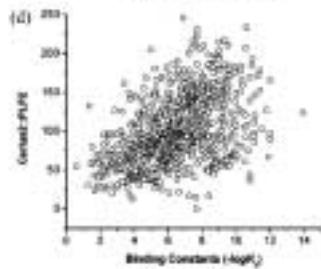
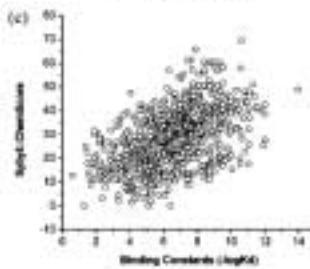


Performance of Different Scoring Functions

(n = 800)

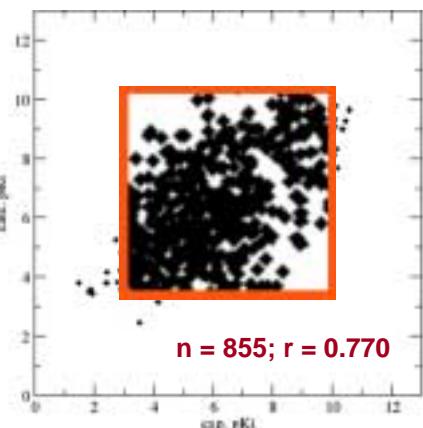
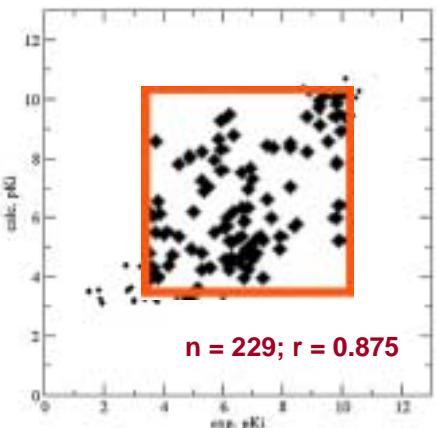


- a) X-Score
- b) DrugScore
- c) ChemScore
- d) PLP2



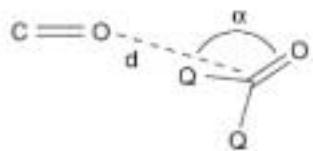
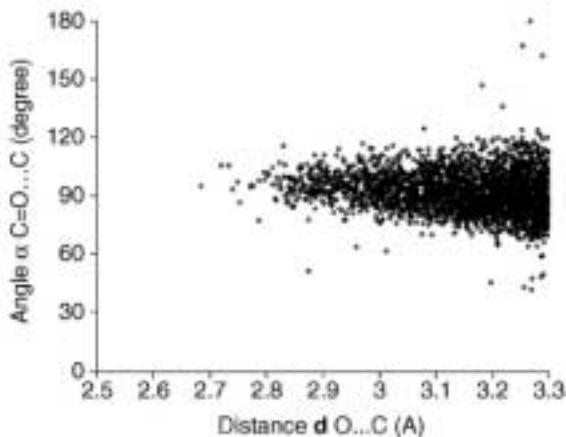
R. Wang et al.,
J. Chem. Inf.
Model. **44**, 2114-
2125 (2004)

SFCscore (Scoring Function Consortium): Affinity Prediction of Protein-Ligand Complexes



C. A. Sottriffer et al., Proteins **73**, 395-419 (2008); cf. A. M. Davis
et al., Angew. Chem. Int. Ed. Engl. **42**, 2718-36 (2003)

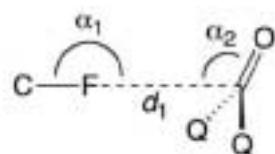
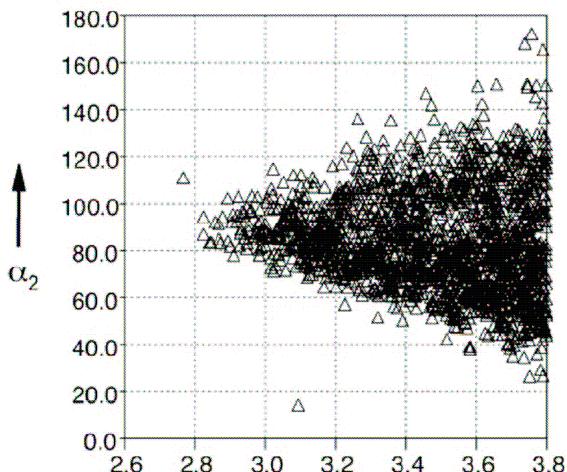
Unrecognized Favorable Interactions



derived from 2,850
high-resolution CSD
structures ($Q = C, N, O$)

T. Schulz-Gasch and M. Stahl, Drug Discov.
Today: Technologies 1, 231-239 (2004)

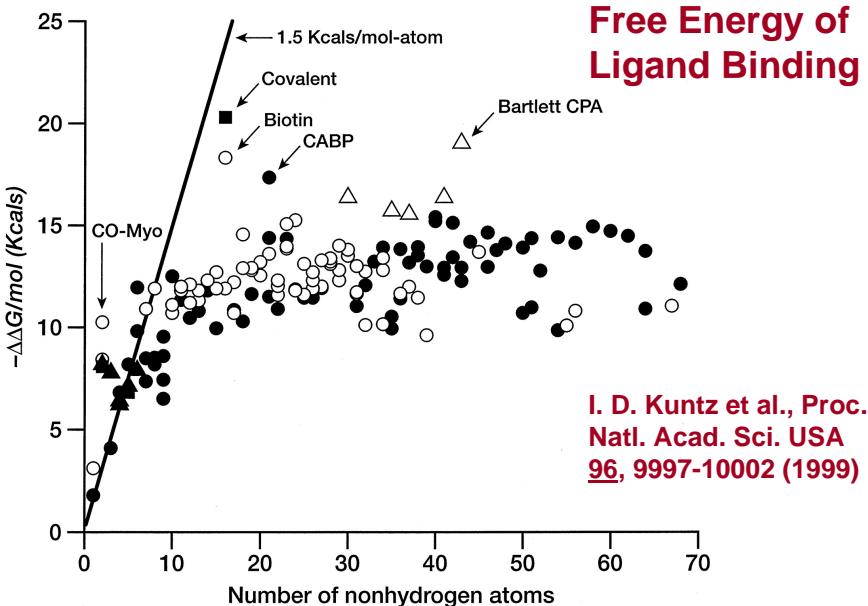
Unrecognized Favorable Interactions



derived from 1,087
high-resolution CSD
structures ($Q = C, N, O$)

M. Zürcher and F. Diederich, J.Org. Chem. 73, 4345-4361 (2008)

Free Energy of Ligand Binding



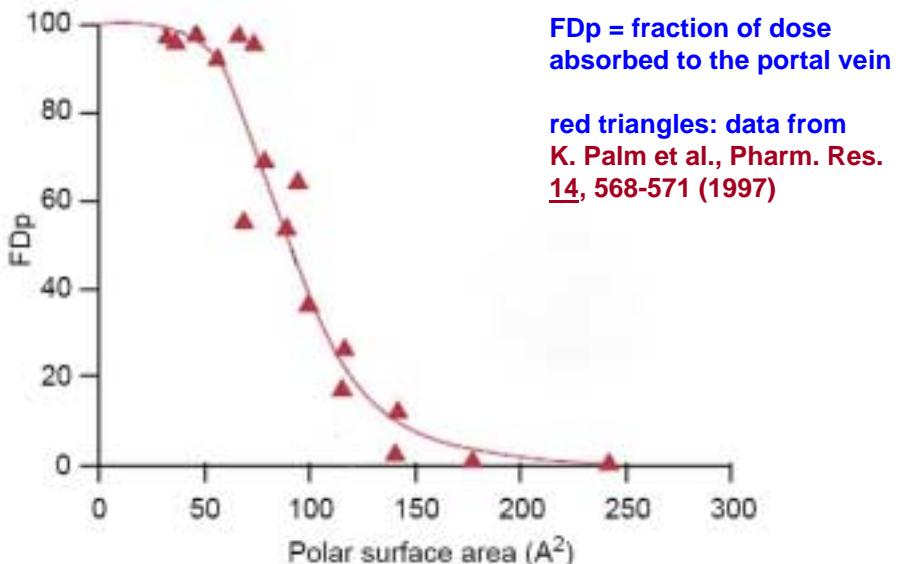
Factors to be Considered in Scoring Functions

- Desolvation enthalpy and entropy (ligand and protein)
- Protonation state of the ligand and the binding site
- Distortion energy of the ligand and its binding site
- Loss of translational and rotational degrees of freedom of the ligand
- MEP + dielectric constant at the binding site
- Dipole moment of the ligand and local dipole moment at the binding site
- Binding enthalpy of the ligand-protein complex
- Repulsive effects (e.g. $-O \cdots O-$)
- Inserted water molecules
- Solvation enthalpy and entropy of the complex

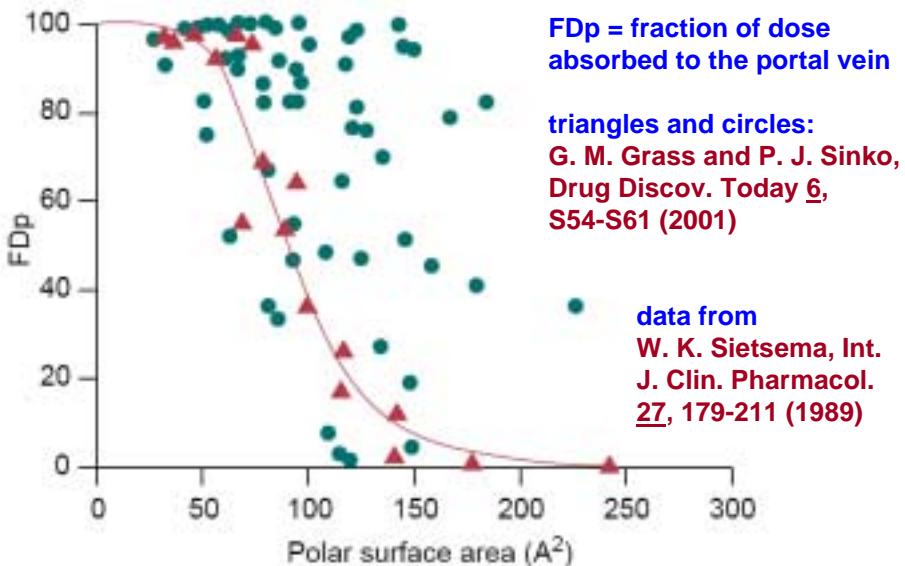
Drug Discovery Bottlenecks of the Past

| Problem | Solution |
|--------------------------|--|
| Target search | genome information |
| Target validation | knock-outs, RNA silencing |
| Lead search | in vitro test models, HTS, VS |
| Lead optimization | automated parallel syntheses, chemogenomics |
| Absorption, permeability | Lipinski rules, Caco cells, formulation, prodrugs |
| Metabolism | MetaSite, MetaPrint2D, liver microsomes, hepatocytes |
| Toxicity | Ames test, hERG models, etc. |
| Drug-drug interactions | CYP inhibition/induction |

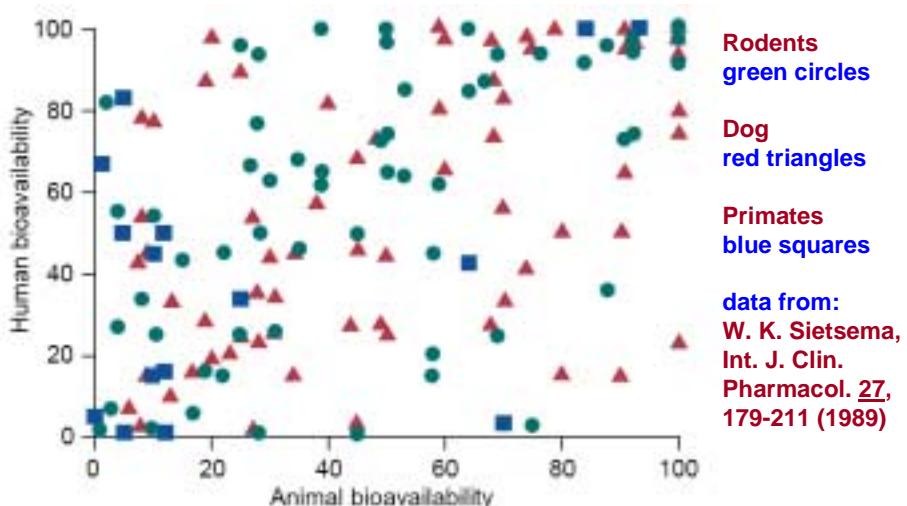
Human Absorption and Polar Surface Area



Human Absorption and Polar Surface Area

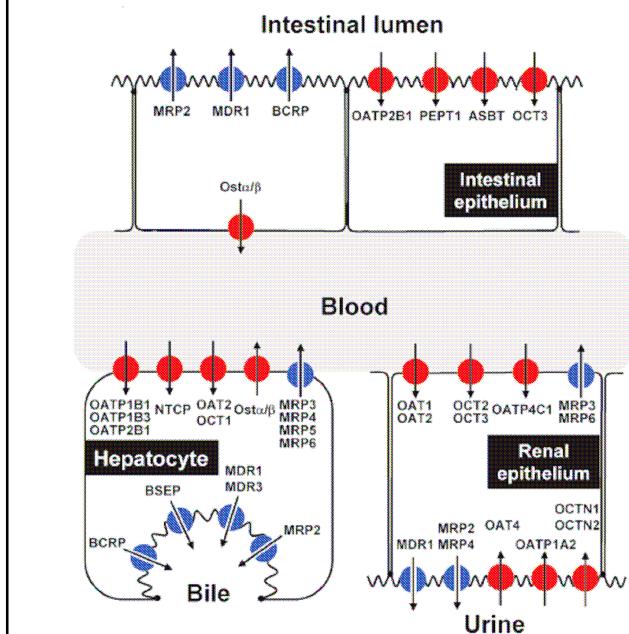


Rodent, Dog, Primate and Human Bioavailability



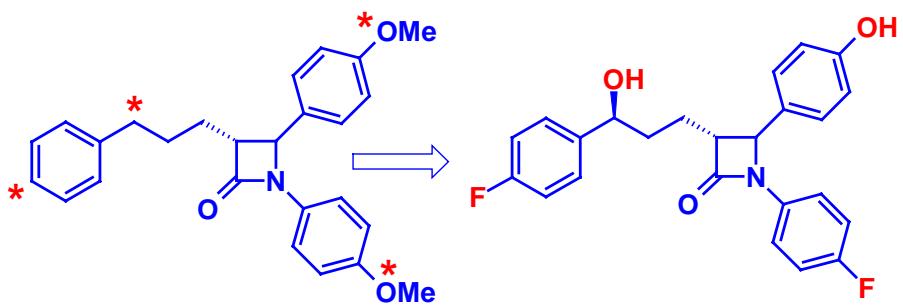
G. M. Grass and P. J. Sinko, Drug Discov. Today 6, S54-S61 (2001)

The Role of Transporters in Drug Absorption and Elimination



H. Gleaser et al.,
in R. J. Vaz and
T. Klabunde,
Antitargets,
Wiley-VCH, 2008,
pp. 341-366

Oxidative Metabolism and Drug Design



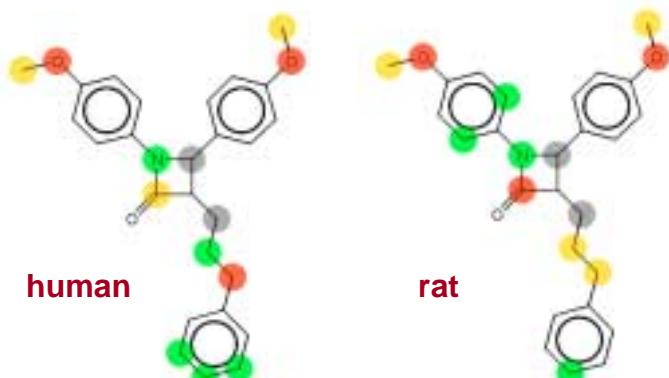
SCH 48461
 ED_{50} (hamster) = 2.2 mg/kg

Ezetimib (SCH 58235, oral cholesterol absorption inhibitor)
 ED_{50} (hamster) = 0.04 mg/kg

M. van Heek et al., J. Pharmacol. Exp. Ther. 283, 157-163 (1997);
D. A. Smith, H. van de Waterbeemd and D. K. Walker, Pharmacokinetics and Metabolism in Drug Design, Wiley-VCH, 2001, p. 85

Prediction of Drug Metabolism: MetaPrint2D

predictions
for human,
dog, rat, all



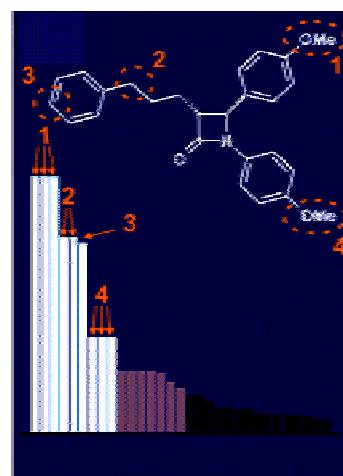
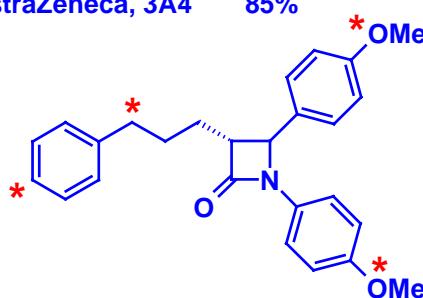
red = high probability
orange = medium probability
green = low probability
white = no probability

S. Boyer et al.,
www-metaprint2d.ch.cam.ac.uk/

Prediction of Drug Metabolism: MetaSite

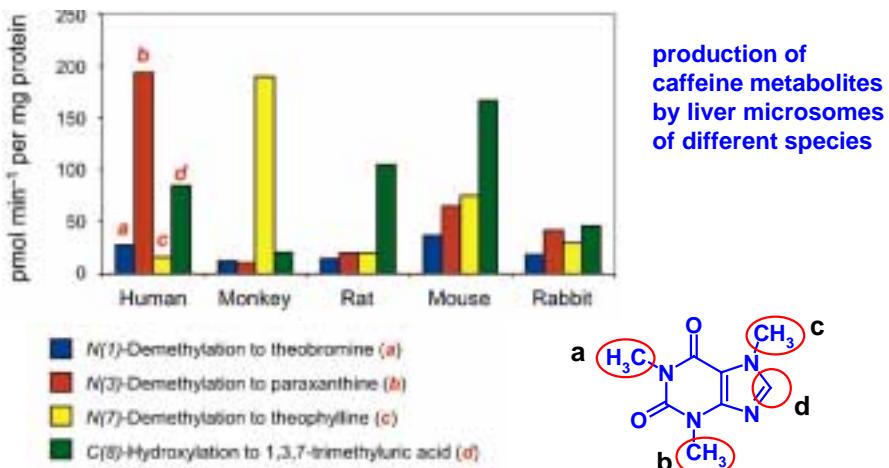
correct predictions:

| | |
|---------------------|-----|
| Sanofi-Aventis, 2C9 | 84% |
| Pfizer, 2D6 | 85% |
| 3A4 | 86% |
| J&J, 2C9, 2D6, 3A3 | 85% |
| AstraZeneca, 3A4 | 85% |



G. Cruciani et al., J. Med. Chem. 48, 6970-6979 (2005)

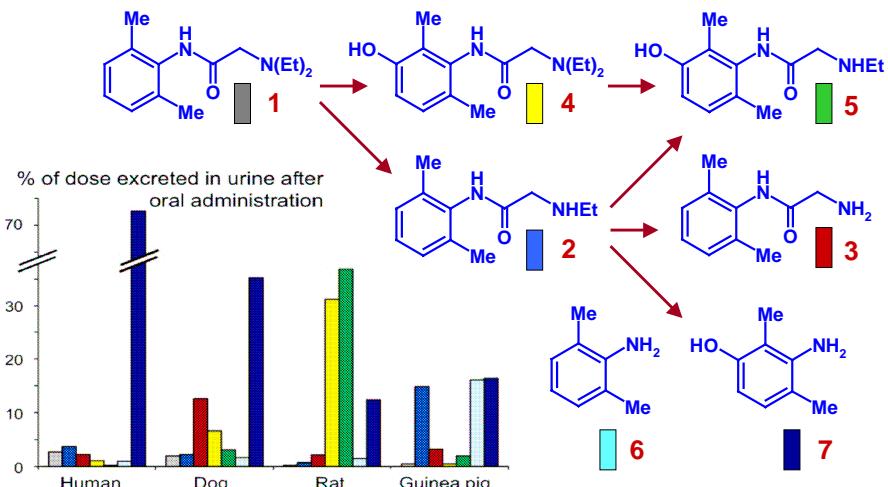
Species Differences of Caffeine Metabolism



F. Berthou et al., Xenobiotica 22, 671-680 (1992)

figure: S. D. Krämer and B. Testa, Chemistry & Biodiversity 5, 2465-2578 (2008)

Species Differences of Lidocaine Metabolism



J. B. Keenaghan and R. N. Boyes, J. Pharmacol. Exp. Ther. 180, 459-463 (1972)

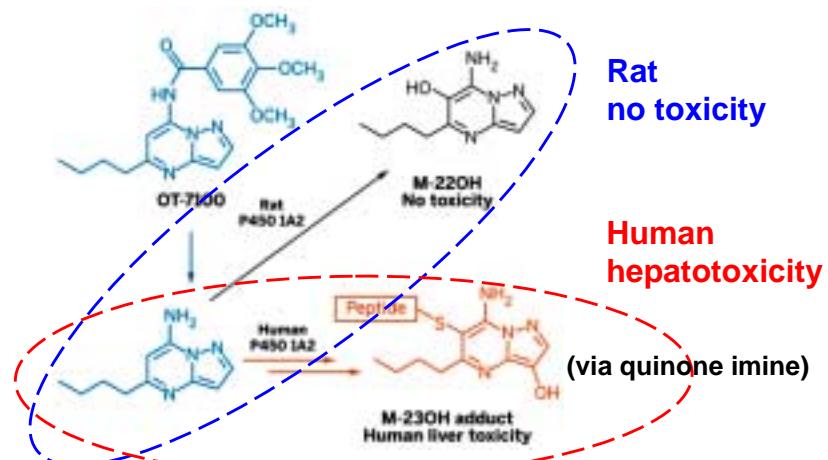
figure: S. D. Krämer and B. Testa, Chemistry & Biodiversity 5, 2465-2578 (2008)

Biological Activities of Metabolites

| Compound | monoamine uptake inhibition rat synaptosomes, IC ₅₀ in nM | | |
|------------------------|---|-----|------|
| Sibutramine (racemate) | DAT | NET | SERT |
| | 1200 | 350 | 2800 |
| (R) NHMe | 12 | 4 | 44 |
| (S) NHMe | 180 | 870 | 9200 |
| (R) NH ₂ | 9 | 13 | 140 |
| (S) NH ₂ | 12 | 62 | 4300 |

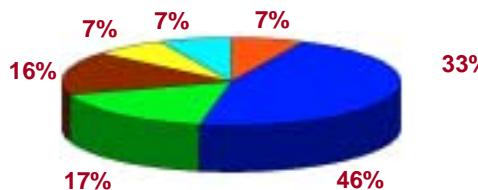
D. L. Nelson and D. R. Gehlert, Endocrine **29**, 49-60 (2006);
data from S. D. Glick et al., Eur. J. Pharmacol. **397**, 93-102 (2000)

Biological Activities of Metabolites



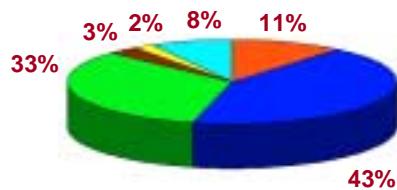
S. Kurabayashi et al., Chem Res. Toxicol. **22**, 323-331 (2009);
cf. Chem. & Eng. News, August 31, 2009, p. 27

Reasons for Failure in Drug Development



- ADME
- Lack of efficacy
- Animal toxicity
- Adverse effects in man
- Commercial reasons
- Miscellaneous

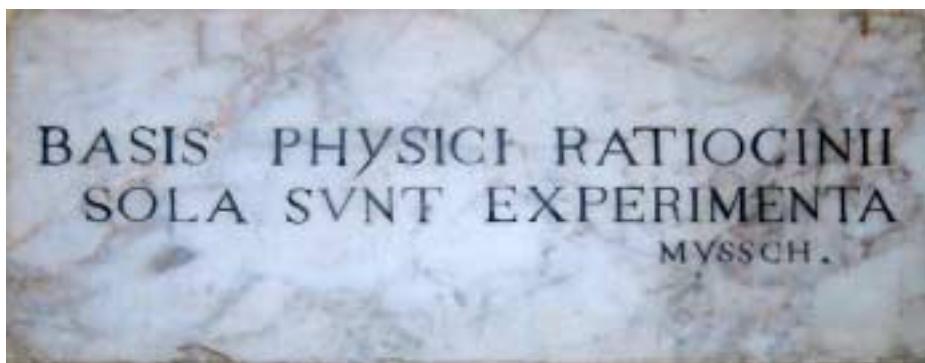
Reasons for failure in clinical development, 1964-1985
(n = 121; without antiinfectives)
T. Kennedy, Drug Discov. today 2, 436-444 (1997)



- Liberation + ADME
- Lack of efficacy
- Toxicity
- Economic
- Other
- Not published

Reasons for failure in clinical development, 1992-2002 (n = 73)
(reasons for market withdrawal, n = 16: toxicity 93%, efficacy 7%)
D. Schuster et al., Curr. Pharm. Design 11, 3545-3559 (2005)

Thank you



Pieter van Musschenbroek (1692-1761)
Tentamina Experimentorum Naturalium
(Museo di Storia Naturale dell'Accademia dei Fisiocritici di Siena)